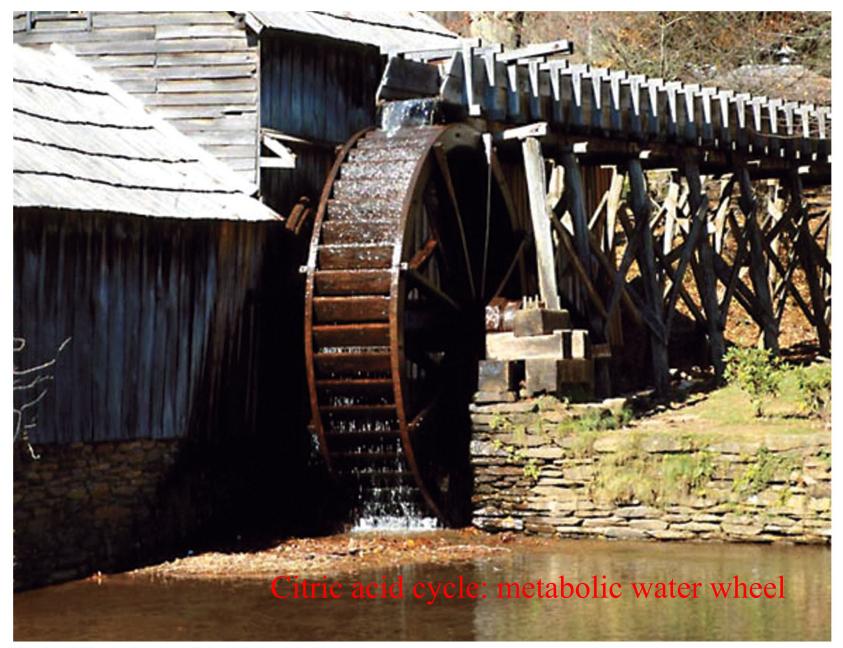
**Chapter 4-I:** Citric Acid Cycle

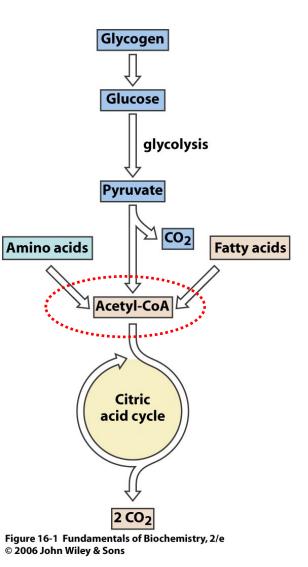


Chapter 16 Opener Fundamentals of Biochemistry, 2/e

## Overview of metabolic fuel metabolism

Citric acid cycle (CAC)

Not merely an oxidation of pyruvate to  $CO_2$ A central pathway for recovering energy from several metabolic fuels



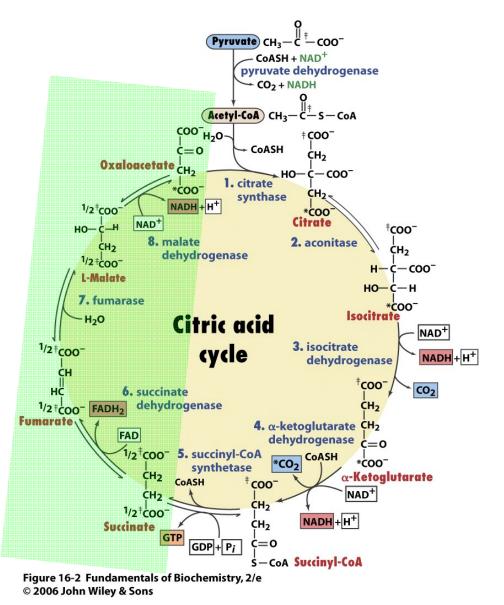
# Overview of CAC

8 reactions oxidizing acetyl-CoA to 2 CO<sub>2</sub> Generation of 3 NADH, 1 FADH2, 1 ATP(GTP)

1930s Hans Krebs linked the already known compounds <u>Plant products</u> Citric acid: citrus fruit Aconitate: monkshood Succinate: amber Fumarate: Fumaria Malate: apple

#### **General features**

Circular pathway: TCA cycle Net reaction Mitochondrial location Amphibolic pathway: provides biosynthetic intermediates



# Synthesis of Acetyl-CoA

Pyruvate to acetyl-CoA (high-E compound) Transport to mitochondria via pyruvate-H<sup>+</sup> symport

Pyruvate dehydrogenase multienzyme complex (PDH) Noncovalently associated enzymes

E. coli enzyme

Pyruvae dehydrogenase (E1): 24 subunits Dihydrolipoyl transacetylase (E2): 24 subunits Dihydrolipoyl dehydrogenase (E3): 12 subunits

Multienzyme complex: evolution of catalytic efficiency

- 1. Overcome diffusion
- 2. Minimize side reactions
- 3. Coordinate control

EM of *E. coli* pyruvate dehydrogenase complex

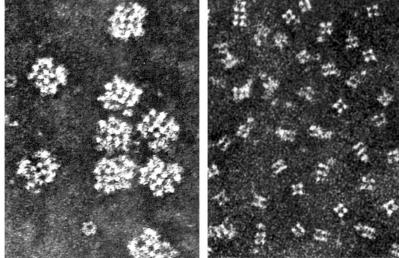


Figure 16-3a Fundamentals of Biochemistry, 2/e

Figure 16-3b Fundamentals of Biochemistry, 2/e

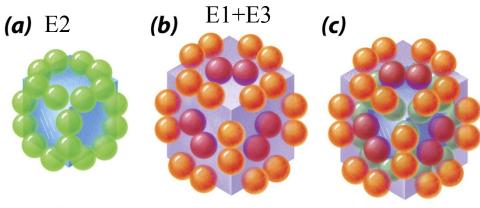
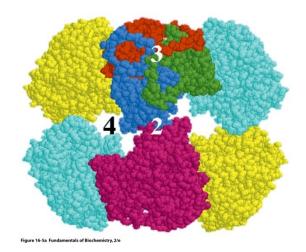


Figure 16-4 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

Different complexity of PDH in different species

E2 core: 24 subunits (8 trimers)



E2 core: 60 subunits (20 trimers)

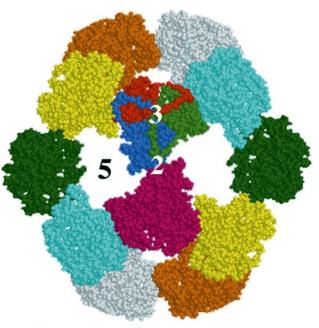


Figure 16-5b Fundamentals of Biochemistry, 2/e

#### The reactions of PDH

 $Pyruvate + CoA + NAD^{+} \rightarrow acetyl-CoA + CO_{2} + NADH + H^{+}$ 

#### 5 different coenzymes TPP

Lipoic acid CoA FAD NAD<sup>+</sup>

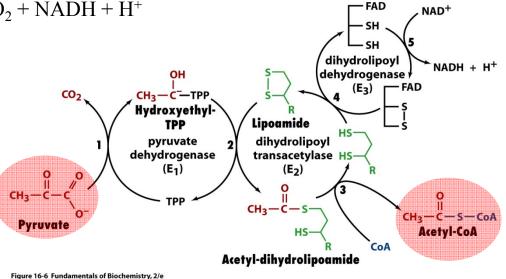


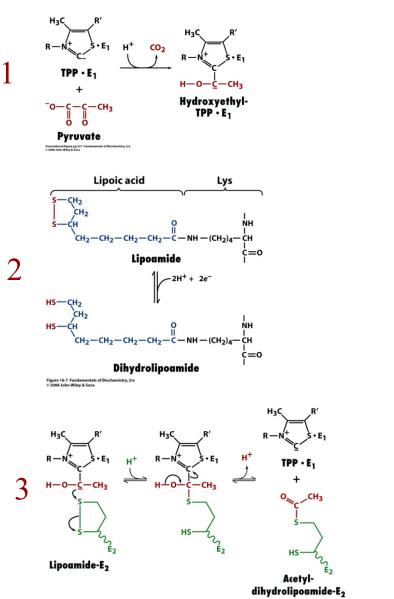
Figure 16-6 Fundamentals of Biochemistry, 2/ © 2006 John Wiley & Sons

Table 16-1 The Coenzymes and Prosthetic Groups of Pyruvate Dehydro
--

Cofactor	Location	Function
Thiamine pyrophosphate	Bound to E <sub>1</sub>	Decarboxylates pyruvate yielding a hydroxyethyl-
(TPP)		TPP carbanion
Lipoic acid	Covalently linked to a Lys on E <sub>2</sub> (lipoamide)	Accepts the hydroxyethyl carbanion from TPP as an acetyl group
Coenzyme A (CoA)	Substrate for $E_2$	Accepts the acetyl group from lipoamide
Flavin adenine dinucleotide (FAD)	Bound to E <sub>3</sub>	Reduced by lipoamide
Nicotinamide adenine dinucleotide (NAD <sup>+</sup> )	Substrate for $E_3$	Reduced by FADH <sub>2</sub>

Table 16-1 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

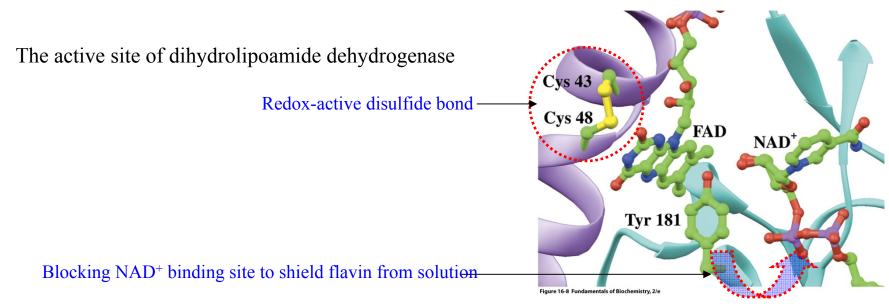
#### The sequence of reactions



0 CoA-S-C-CH3 Acetyl-CoA CH<sub>3</sub> + HS CoA-SH 4 HS-HS-Z E2 SC E2 Acetyl-Dihydrolipoamide-E<sub>2</sub> dihydrolipoamide-E<sub>2</sub> Unnumbered figure pg 523b F © 2006 John Wiley & Sons -FAD HS-FAD +  $\longrightarrow$ SH + S 5 HS SH E<sub>3</sub> (oxidized) E2 E<sub>3</sub> (reduced) É2 Unnumbered figure pg 523c Fundar © 2006 John Wiley & Sons entals of Biochemistry, 2/e  $NAD^+ NADH^+H^+$ FADH2 FAD -FAD - SH S 6 SH S E<sub>3</sub> (oxidized)

Unnumbered figure pg 523d Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

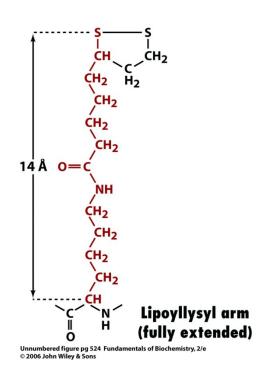
Unnumbered figure pg 523a Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons



Check reduction potential of FAD &  $NAD^+$ 

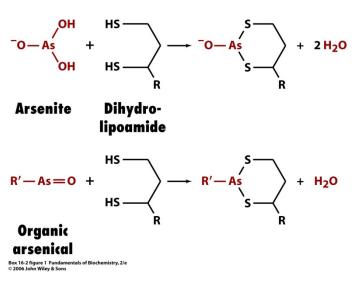
The lipoamide group of E2

Channeling of reaction intermediates between E2 (inside) and E1+E3 (outside) Swing action from E1 to the E2 active site, and then to E3 Flexibility of the bound domain

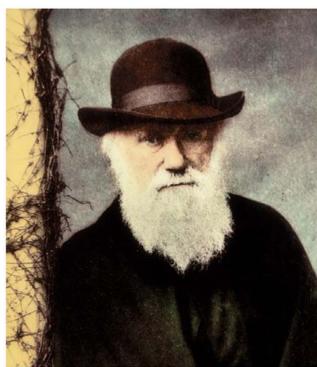


## Arsenic poisoning

Arsenite Organic arsenicals Binding to -SH of lipoamide Inactivation of PDH & α-ketoglutarate dehydrogenase







Box 16-2 figure 2 Fundamentals of Biochemistry, 2/e

Box 16-2 figure 3 Fundamentals of Biochemistry, 2/e

## Enzymes of the citric acid cycle

 $\frac{8 \text{ enzymes}}{\text{Citrate synthase}}$ Citrate synthase
Aconitase
NAD<sup>+</sup>-dependent isocitrate dehydrogenase: CO<sub>2</sub> release  $\alpha$ -ketoglutarate dehydrogenase: CO<sub>2</sub> release
Succinyl-CoA synthetase: GTP generation
complete oxidation of acetyl-CoA

Succinate dehydrogenase Fumarase Malate dehydrogenase

Regeneration of oxaloacetate

#### Citrate synthase

Condensation of acetyl-CoA and oxaloacetate Ordered sequential reaction oxaloacetate followed by acetyl-CoA

Large conformational changes sealing the oxaloacetate binding site generating acetyl-CoA binding site

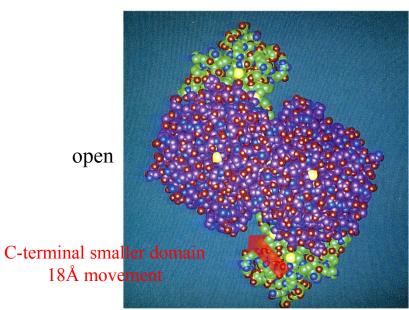
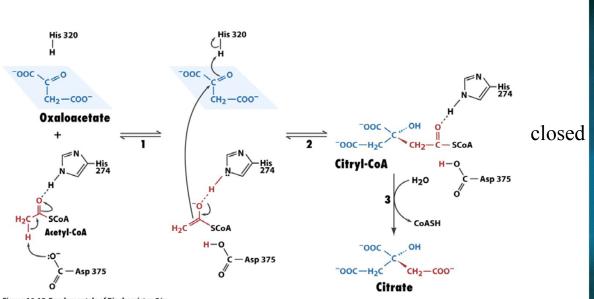


Figure 16-9a Fundamentals of Biochemistry, 2/e



Fjure 15-9F Fundamentals of Biochemistry. 2/e

Reaction mechanism

Figure 16-10 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

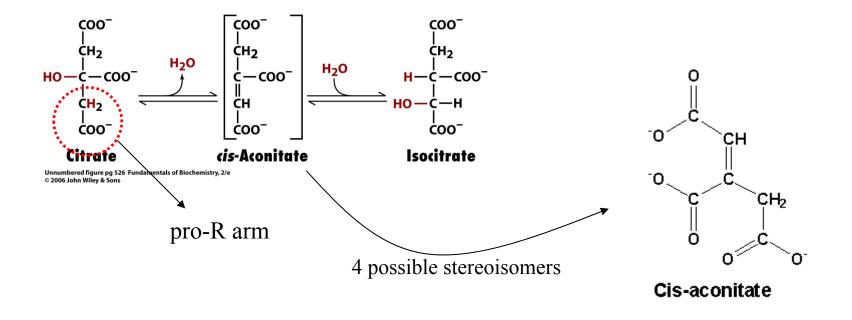
#### <u>Aconitase</u>

Reversible isomerization of citrate and isocitrate with cis-aconitate as an intermediate Dehydration and rehydration

Citrate: pro-chiral compound

Isocitrate: chiral compound

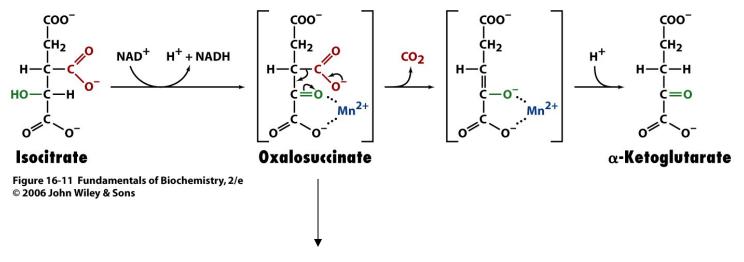
[4Fe-4S] iron-sulfur cluster: (normally involve in redox reactions)



## NAD<sup>+</sup>-dependent isocitrate dehydrogenase

There is a mammalian isozyme that depends on NADP<sup>+</sup> (cytosolic and unrelated to citric acid cycle) Oxidative decarboxylation (similar to phosphogluconate dehydrogenase in PPP)

oxidation of secondary alcohol to ketone decarboxylation of  $\beta$ -carboxyl group polarization of carbonyl group by  $Mn^{2+}$  (or  $Mg^{2+}$ ) first  $CO_2$  generation

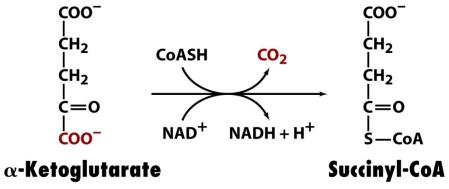


Transient ( $\beta$ -keto acid is unstable)

#### $\alpha$ -ketoglutarate dehydrogenase

Oxidative decarboxylation of  $\alpha$ -keto acid Second CO<sub>2</sub> generation Chemically resemble the reaction by PDH

Multienzyme complex α-ketoglutarate dehydrogenase (E1) Dihydrolipoyl transsuccinylase (E2) Dihydrolipoyl dehydrogenase (E3): identical to the E3 of PDH



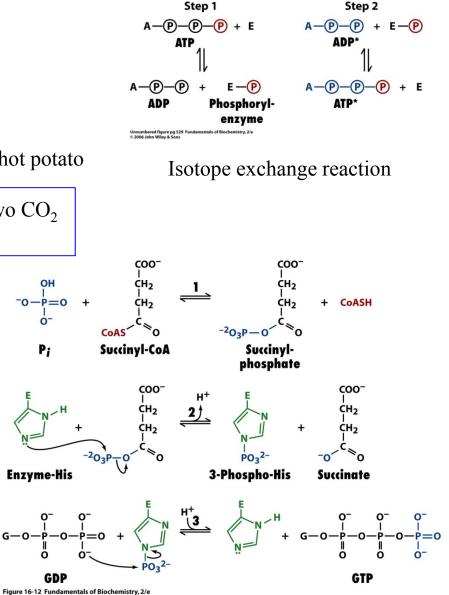
Unnumbered figure pg 528 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

#### Succinyl-CoA synthetase

Succinate thiokinase Generation of GTP (high-E succinyl-CoA): substrate level phosphorylation How?

Phosphoryl enzyme intermediate: passing of a hot potato

One acetyl equivalent is completely oxidized to two  $CO_2$ Generation of two NADH + one GTP

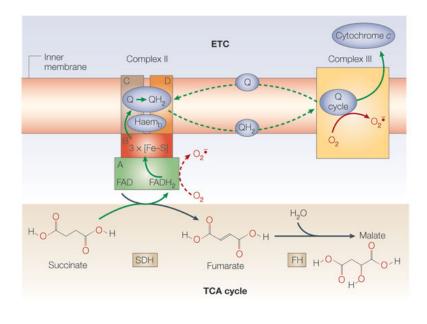


© 2006 John Wiley & Sons

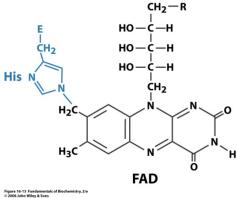
#### Succinate dehydrogenase

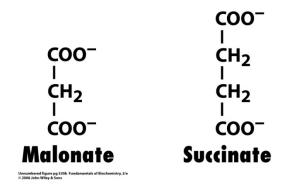
Stereospecific dehydrogenation of succinate (alkane) to fumarate (alkene) Inhibited by malonate FAD prosthetic group

Only membrane bound enzyme to funnel electrons to ETS



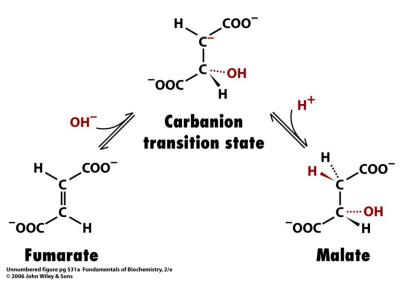






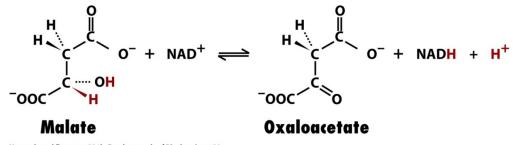
#### Fumarase (fumarate hydratase)

Hydration of the double bond to form malate via carbanion transition state



#### Malate dehydrogenase

Regeneration of oxaloacetate The same catalytic reaction with ADH & LDH



Unnumbered figure pg 531b Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons