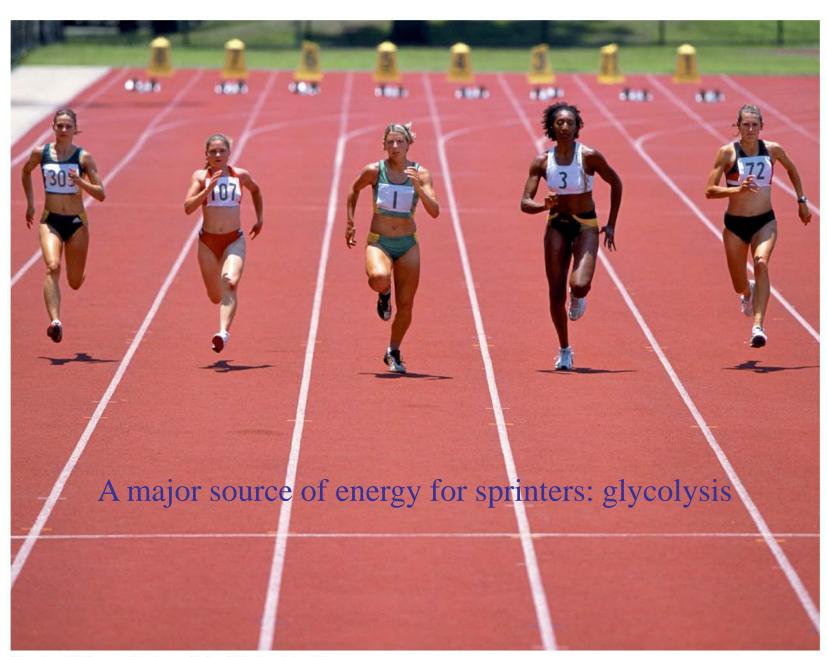
# **Chapter 2-I:** Glucose Catabolism



Chapter 14 Opener Fundamentals of Biochemistry, 2/e

Glycolysis Embden-Meyerhof-Parnas pathway

Generation of two pyruvate molecules 2 ATP 2 NADH

10 enzymatic reactions to generate high-E compounds

stage I: two glyceraldehyde-3-P two ADP stage II: two pyruvate four ATP

Bypass to pentose phosphate pathway

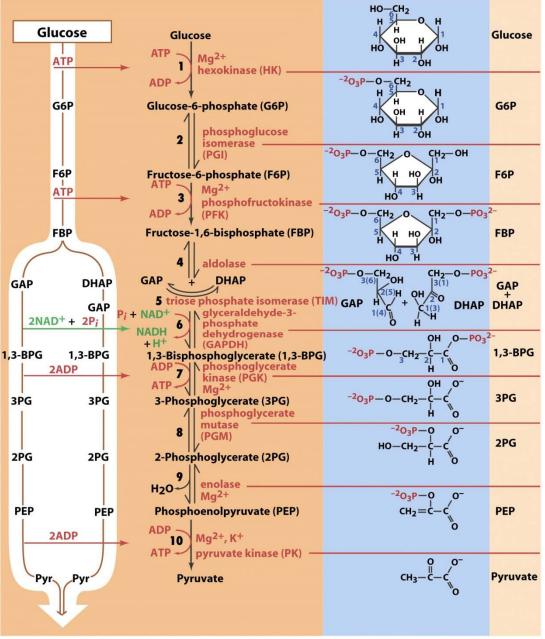
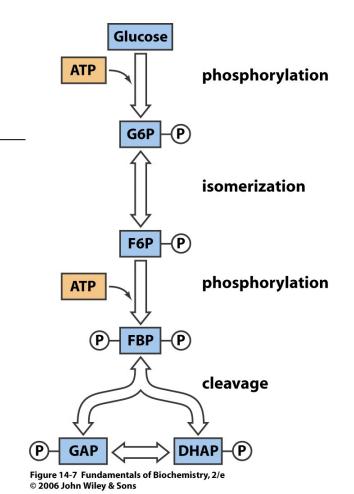


Figure 14-1 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

## Stage I

Hexokinase: glucose to G6P (ATP to ADP) Phosphoglucose isomerase (PGI): G6P to F6P Phosphofructokinase (PFK): F6P to FBP (ATP to ADP) Aldolase: FBP to DHAP & GAP Triose phosphate isomerase (TIM): DHAP to GAP

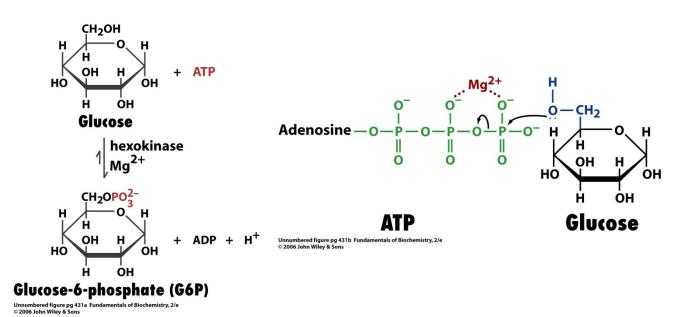
Net: glucose to 2 GAP (2ATP to 2ADP)



### Hexokinase

Nonspecific enzyme low Km

Glucokinase Liver enzyme high Km blood glucose control



Substrate induced conformational change prevent ATP hydrolysis

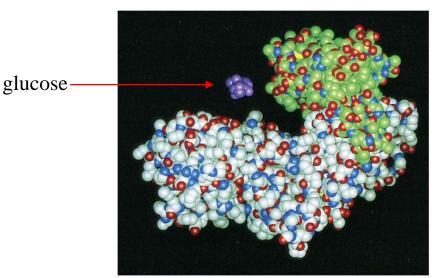


Figure 14-2a Fundamentals of Biochemistry, 2/e

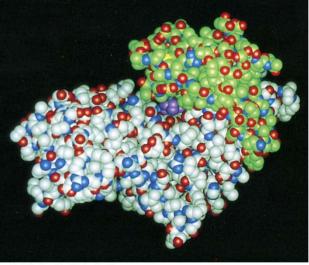
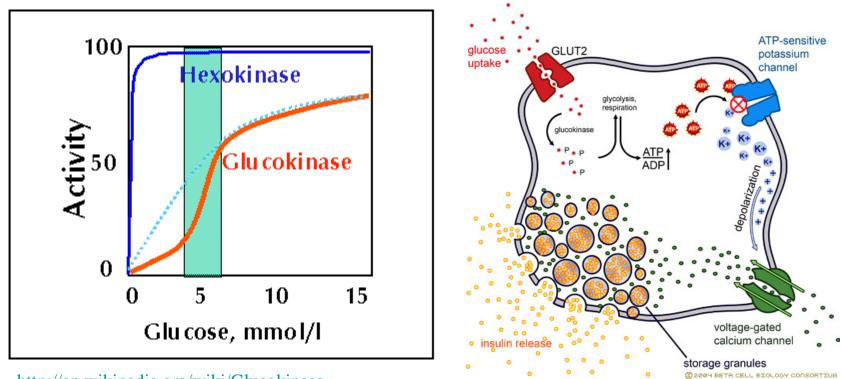


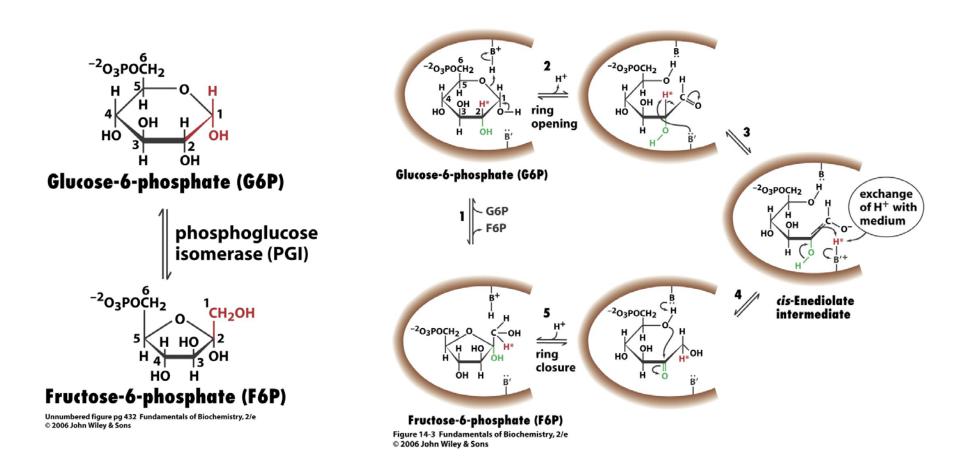
Figure 14-2b Fundamentals of Biochemistry, 2/e

### A glucose sensor in beta cells



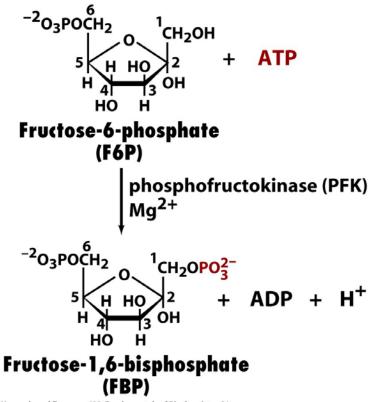
http://en.wikipedia.org/wiki/Glucokinase

## Phosphoglucose isomerase (PGI)



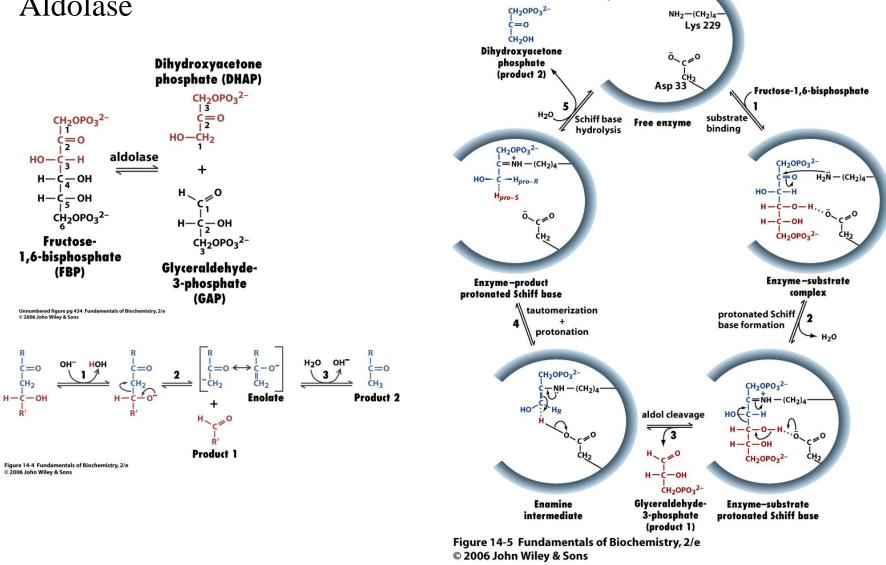
### Phosphofructokinase (PFK)

Central role in control of glycolysis as a rate-determining step Allosteric regulation



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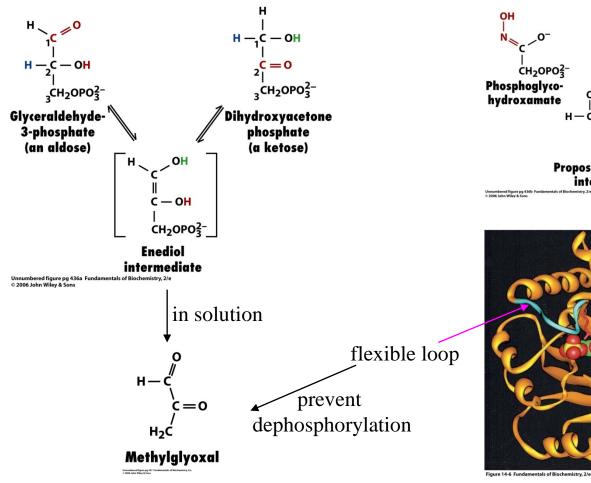
### Aldolase

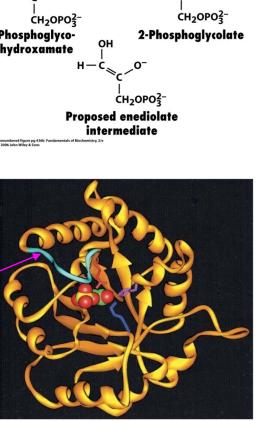


Stabilization of enolate intermediate through increased electron delocalization

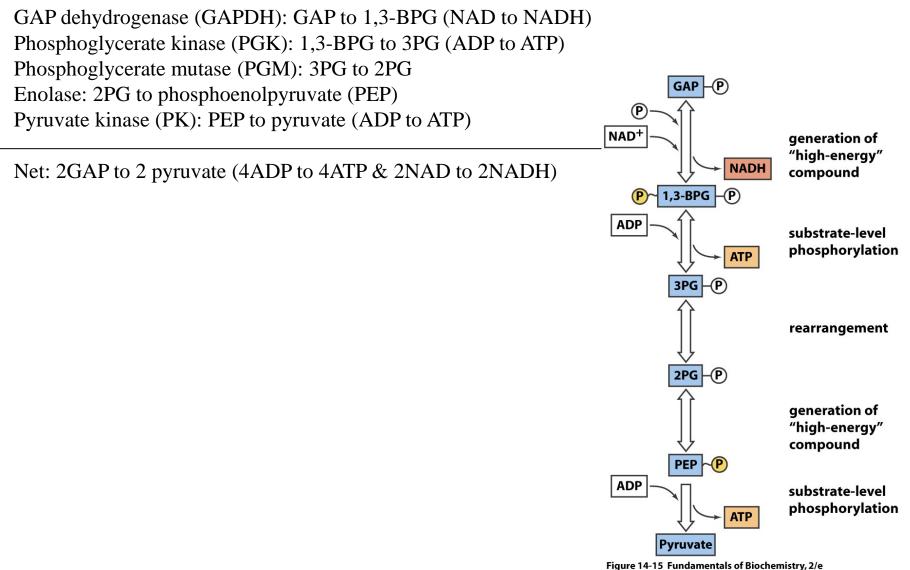
## Triose phosphate isomerase (TIM)

DHAP-GAP: ketose-aldose isomers Isomerization via enediol intermediate General acid-base catalysis Catalytically perfect enzyme: diffusion controlled reaction rate  $Keq = [GAP]/[DHAP] = 4.73 \times 10^{-2}$ 





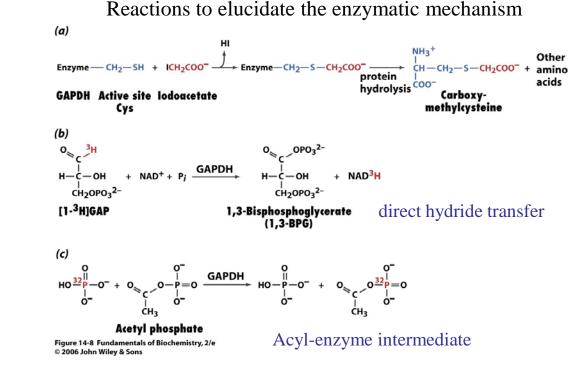
## Stage II

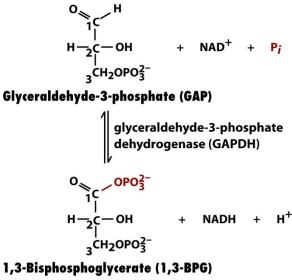


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### GAP dehydrogenase (GAPDH)

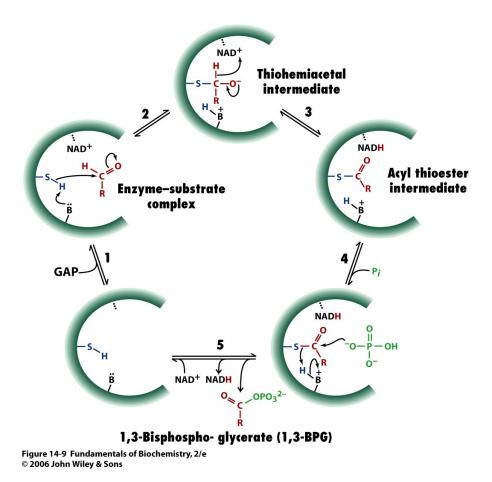
Generation of high-E compound Driven by oxidation of aldehyde Generation of NADH Slightly unfavorable reaction:  $\Delta G^{o'} = +6.7$  kJ/mol





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## Enzymatic mechanism of GAPDH



## Phosphoglycerate kinase (PKG)

First ATP generation: substrate level phosphorylation Strong exergonic reaction: coupled to GAPDH rxn to make ATP and NADH

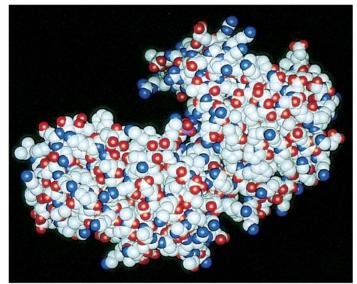
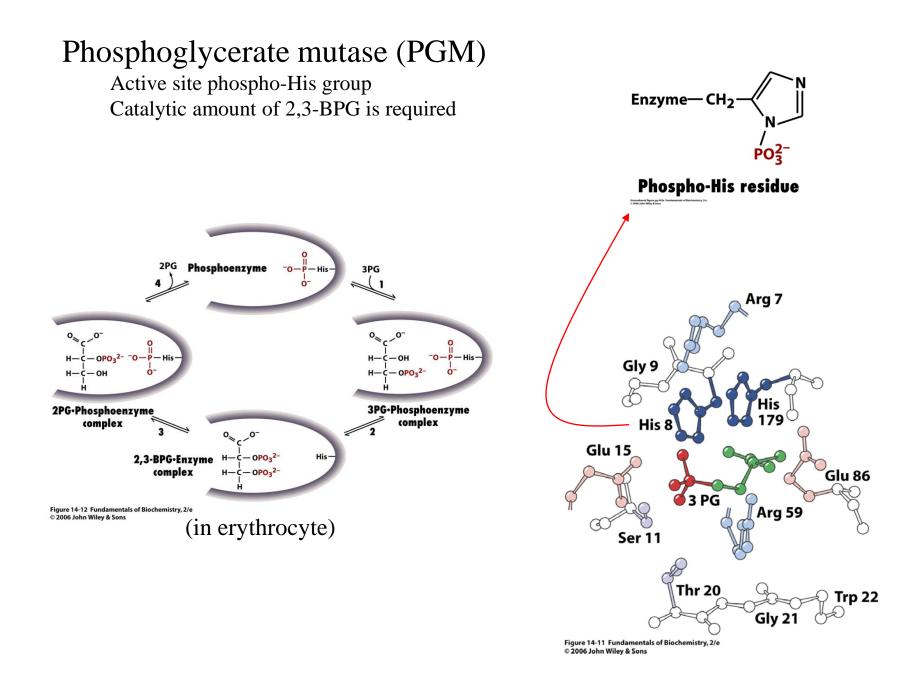


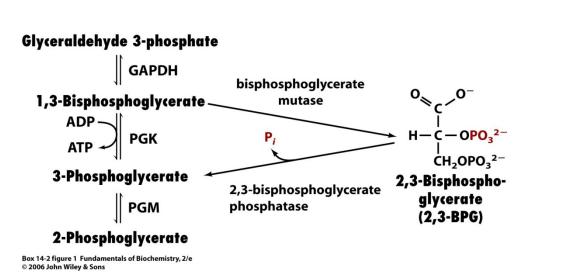
Figure 14-10 Fundamentals of Biochemistry, 2/e

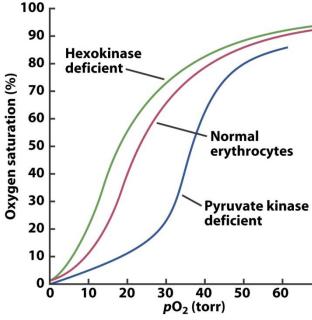
#### 3-Phosphoglycerate (3PG)

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### 2,3-BPG in erythrocyte



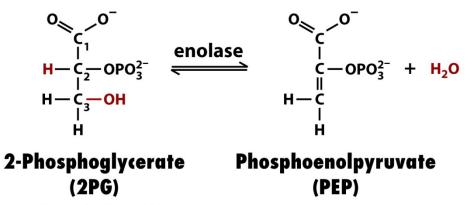


Box 14-2 figure 2 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

Explain the figure

### Enolase

Generation of 2<sup>nd</sup> high-E intermediate Dehydration rxn to produce unstable enol form 2PG hydrolysis release insufficient free E



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## Pyruvate kinase (PK)

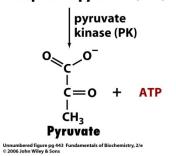
Conversion of unstable enol to keto form Generation of ATP

$$O_{C} O^{-}$$

$$C_{C} O^{-} O^{-}$$

$$C_{C} O^{-} O^{-}$$

Phosphoenolpyruvate (PEP)



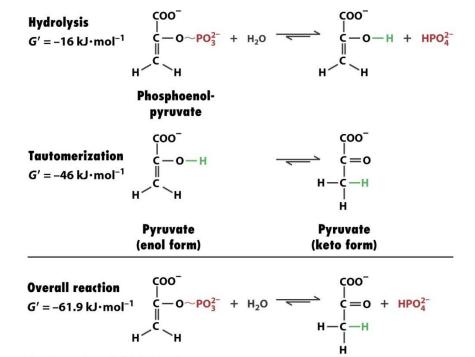


Figure 14-14 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

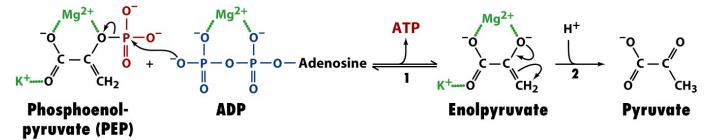
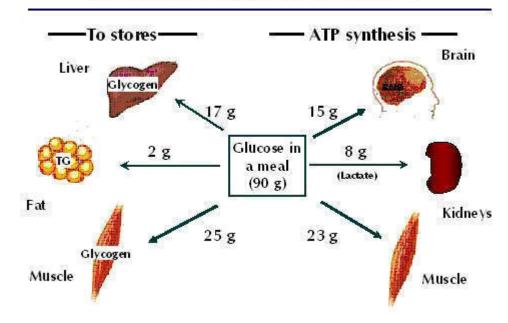


Figure 14-13 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

3 glycolytic products

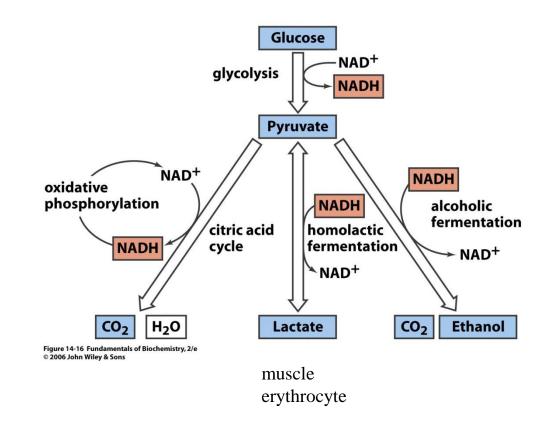
ATP NADH: electron transport Pyruvate: fermentation



#### Distribution of glucose after a meal

Fermentation: the anaerobic fate of pyruvate

Aerobic condition: pyruvate to citric acid cycle Anaerobic condition: lactate or alcohol fermentation reduction of pyruvate regeneration of NAD+

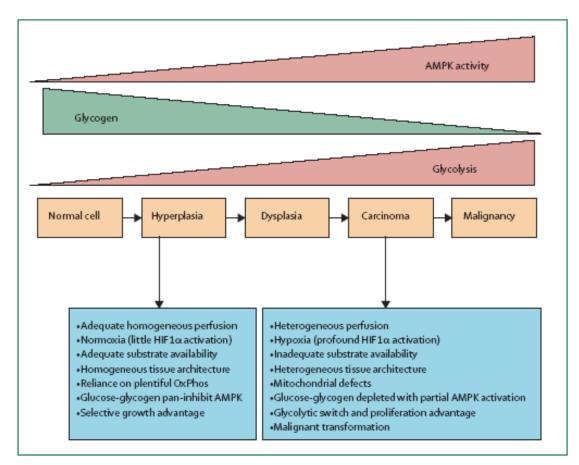


## Otto Warburg (1931)

Cancer cells have increased glycolysis and impaired OXPHOS

Tumor cell glycolysis >>> normal cells (~80% of glucose)

AMPK (AMP-activated protein kinase) senses AMP/ATP ratio drives glycolysis via HK, GLUT1 induction



### Alcoholic fermentation

Pryuvate to ethanol and CO<sub>2</sub>

Two consecutive reactions via acetaldehyde

Pyruvate decarboxylase: TPP (thiamine pyrophosphate) as a coenzyme

Alcohol dehydrogenase (ADH):

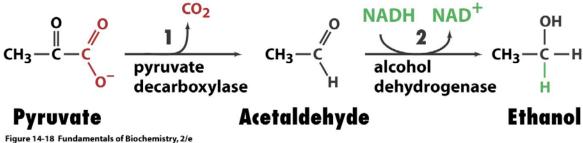
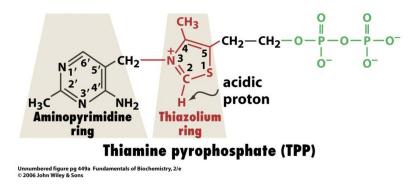


Figure 14-18 Fundamentals of Biochemistry, 2 © 2006 John Wiley & Sons



### Thiamine (vitamine B1) deficiency

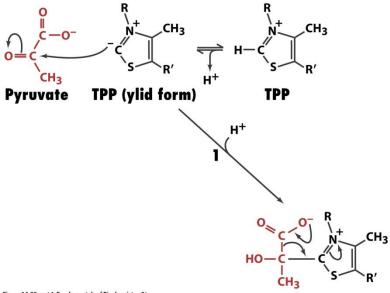
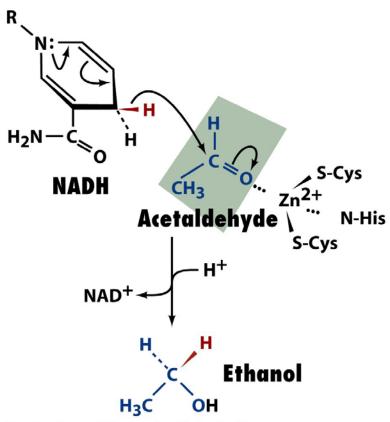


Figure 14-20 part 1 Fundamentals of Biochemistry, 2/e © 2006 John Wiley & Sons

Electron sink

### ADH

Zn enzyme Stabilize the developing negative charge in the transition state Acetaldehyde to ethanol and the reverse



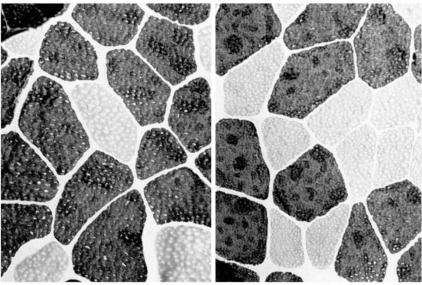
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Glycolysis in skeletal muscles

Slow-twitching (type I) rich in mitochondria (red fiber: heme-containing cytochrome)

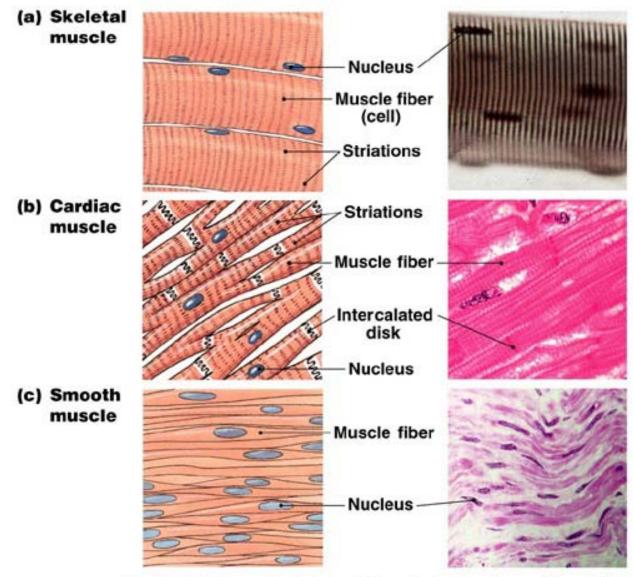
Fast-twitching (type II) devoid of mitochondria (white fiber)

How about birds? How about sprinters and distance runners?



Slow-twitch muscle fiber Box 14-3 Fundamentals of Biochemistry, 2/e

Fast-twitch muscle fiber



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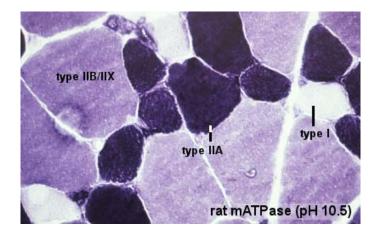
Fig. 12-1

### Comparison of Slow and Fast Twitch Fibers

Use (examples) Motor unit size ATPase activity\* Contraction speed Fatigue resistance Myoglobin content Capillary density Fiber color Glycolytic enzymes Mitochondrial content \*Rapid breakdown of ATP

Type IIa Type IIb Type I (slow-oxidative) (fast-oxidative) (fast-glycolytic) Posture Walking Sprinting 100+ fibers 2-6 fibers 2-6 fibers High High Low Slow Fast Fast High Intermediate Low High High Low High Intermediate Low Red (dark) White Red Low Intermediate High Packed Intermediate Sparse

See also Table 12-2 in Silverthorn



	Type IIa Oxidative	Type IIb Glycolytic
Marathoners	82%	18%
Distance swimmers	74	26
Couch potatoes	45	55
Sprinters	37	63